

Name: _____

Date: _____

Form A

Formula

Name

1	SiO ₂	_____
2	UP	_____
3	XeO ₄	_____
4	BrF	_____
5	ICl	_____
6	NCl ₃	_____
7	AuClO	_____
8	HClO ₄	_____
9	NiH ₂	_____
10	XeF ₆	_____
11	P ₂ I ₄	_____
12	N ₂ F ₄	_____
13	SF ₄	_____
14	PBr ₃	_____
15	Hg ₂ Br ₂	_____
16	CCl ₃	_____
17	H ₃ PO ₃	_____
18	AgHSO ₃	_____
19	Zn(NO ₂) ₂	_____
20	Fe(ClO ₃) ₃	_____
21	LiCN	_____
22	Hg ₂ CO ₃	_____
23	Cu ₃ PO ₄	_____
24	U ₂ S ₅	_____
25	Mn(ClO ₃) ₇	_____
26	Cr(HCO ₃) ₃	_____
27	HBr	_____
28	H ₂ SO ₄	_____
29	SiI ₄	_____
30	HCN	_____
31	Fe ₃ (PO ₃) ₂	_____
32	Ag ₃ PO ₃	_____
33	HNO ₂	_____
34	SiBr ₄	_____
35	NH ₄ H	_____
36	HNO ₃	_____
37	TiI ₄	_____

- 38 AgCl _____
 39 Cd(NO₃)₂ _____
 40 Co₃N₂ _____

Name	Formula
41 bromine trifluoride	_____
42 tungsten (III) hypochlorite	_____
43 chloric acid	_____
44 gold (III) dihydrogen phosphate	_____
45 cesium hypochlorite	_____
46 tin (II) hydroxide	_____
47 manganese (III) bisulfate	_____
48 aluminum sulfate	_____
49 beryllium sulfide	_____
50 sulfurous acid	_____
51 hydroiodic acid	_____
52 tungsten (II) sulfide	_____
53 cobalt (III) phosphate	_____
54 aluminum bicarbonate	_____
55 gold (III) sulfide	_____
56 carbon tetrachloride	_____
57 carbonic acid	_____
58 disulfur dichloride	_____
59 cobalt (III) bisulfite	_____
60 oxalic acid	_____
61 hypochlorous acid	_____
62 lead (II) hypochlorite	_____
63 gold (I) bisulfite	_____
64 cobalt (III) perchlorate	_____
65 xenon difluoride	_____
66 gold (I) chlorate	_____
67 tetraphosphorus trisulfide	_____
68 lead (IV) bromide	_____
69 hydrochloric acid	_____
70 hydrofluoric acid	_____
71 tungsten (V) bisulfate	_____
72 potassium chlorate	_____
73 magnesium bisulfate	_____
74 chlorous acid	_____
75 phosphoric acid	_____
76 chromium (III) dihydrogen phosphate	_____
77 rubidium oxide	_____

- 78 uranium (V) sulfite
- 79 copper (II) fluoride
- 80 hydrosulfuric acid

Name: _____

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Form B

Formula	Name
1 $\text{Ti}(\text{ClO}_4)_4$	_____
2 RbI	_____
3 N_2O_4	_____
4 NiI_2	_____
5 SF_6	_____
6 ClO_2	_____
7 KH	_____
8 $\text{Ra}(\text{NO}_2)_2$	_____
9 $\text{Cr}(\text{OH})_6$	_____
10 SO_2	_____
11 P_2O_5	_____
12 CoI_3	_____
13 XeO_4	_____
14 CdBr_2	_____
15 $\text{W}(\text{ClO})_5$	_____
16 $\text{Be}(\text{ClO})_2$	_____
17 I_2O_5	_____
18 $\text{Cu}(\text{NO}_2)_2$	_____
19 CCl_4	_____
20 FrF	_____
21 $\text{Mn}(\text{NO}_2)_2$	_____
22 $\text{U}(\text{HCO}_3)_3$	_____
23 HCl	_____
24 $\text{Au}(\text{HSO}_4)_3$	_____
25 $\text{Cd}(\text{NO}_3)_2$	_____
26 $\text{Sr}_3(\text{PO}_3)_2$	_____
27 BaI_2	_____
28 $\text{Cu}_3(\text{PO}_3)_2$	_____
29 Sn_3P_2	_____
30 $\text{Au}(\text{ClO}_3)_3$	_____
31 MnF_3	_____
32 XeF_6	_____

33	$\text{Cd}_3(\text{PO}_4)_2$	_____
34	HClO	_____
35	Al_2S_3	_____
36	CH_3COOH	_____
37	AuClO_3	_____
38	S_2Br_2	_____
39	$\text{Ti}(\text{NO}_2)_4$	_____
40	$\text{U}(\text{NO}_2)_3$	_____

	Name	Formula
41	mercury (II) bromide	_____
42	nitrogen triiodide	_____
43	phosphorous acid	_____
44	potassium sulfide	_____
45	carbon dioxide	_____
46	chromium (VI) phosphide	_____
47	hydrobromic acid	_____
48	carbonic acid	_____
49	carbon trichloride	_____
50	lead (IV) phosphate	_____
51	bromine trifluoride	_____
52	oxygen difluoride	_____
53	sulfuric acid	_____
54	sulfur monoxide	_____
55	chlorous acid	_____
56	nitrous acid	_____
57	barium phosphite	_____
58	iron (VI) bisulfite	_____
59	calcium nitrite	_____
60	potassium dihydrogen phosphate	_____
61	xenon trioxide	_____
62	silver fluoride	_____
63	hydrosulfuric acid	_____
64	lead (II) perchlorate	_____
65	nitric acid	_____
66	oxalic acid	_____
67	uranium (IV) perchlorate	_____
68	tungsten (III) oxide	_____
69	rubidium bisulfite	_____
70	hydroiodic acid	_____
71	manganese (II) sulfite	_____
72	sulfurous acid	_____

- | | | |
|----|----------------------------|-------|
| 73 | hydrocyanic acid | _____ |
| 74 | chloric acid | _____ |
| 75 | titanium (III) bicarbonate | _____ |
| 76 | uranium (III) oxide | _____ |
| 77 | beryllium hydroxide | _____ |
| 78 | phosphoric acid | _____ |
| 79 | disulfur decafluoride | _____ |
| 80 | hydrofluoric acid | _____ |

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Form C

Formula	Name
1 XeF ₆	_____
2 Hg(ClO) ₂	_____
3 N ₂ F ₄	_____
4 LiNO ₃	_____
5 RaI ₂	_____
6 KClO	_____
7 AuF ₃	_____
8 PCl ₅	_____
9 RaH ₂	_____
10 ICl	_____
11 NBr ₃	_____
12 CCl ₃	_____
13 LiF	_____
14 HClO ₃	_____
15 NiH ₂	_____
16 Co ₂ S ₃	_____
17 Ag ₃ N	_____
18 Cu(CN) ₂	_____
19 U(HSO ₄) ₃	_____
20 Cr ₂ (SO ₃) ₃	_____
21 Sn(ClO) ₄	_____
22 Co(NO ₃) ₂	_____
23 H ₂ SO ₃	_____
24 Fe(ClO) ₂	_____
25 NCl ₃	_____
26 CuHSO ₄	_____
27 Li ₃ P	_____

28	$\text{Au}(\text{H}_2\text{PO}_4)_3$	_____
29	XeF_2	_____
30	$\text{Ti}(\text{NO}_2)_3$	_____
31	SiF_4	_____
32	$\text{U}(\text{HCO}_3)_4$	_____
33	XeO_3	_____
34	$\text{Sn}(\text{HSO}_3)_2$	_____
35	$\text{Co}(\text{CN})_2$	_____
36	$\text{Pb}(\text{OH})_4$	_____
37	$\text{Mn}(\text{CO}_3)_3$	_____
38	$\text{Mn}(\text{CN})_2$	_____
39	$\text{Sr}_3(\text{PO}_3)_2$	_____
40	HBr	_____

	Name	Formula
41	hydrosulfuric acid	_____
42	gold (I) phosphate	_____
43	bromine pentafluoride	_____
44	tungsten (II) perchlorate	_____
45	hydroiodic acid	_____
46	tetraphosphorus trisulfide	_____
47	acetic acid	_____
48	beryllium chloride	_____
49	cobalt (II) bromide	_____
50	xenon tetroxide	_____
51	phosphorous acid	_____
52	hydrofluoric acid	_____
53	perchloric acid	_____
54	mercury (II) phosphate	_____
55	phosphoric acid	_____
56	rubidium chloride	_____
57	francium iodide	_____
58	nitric acid	_____
59	titanium (II) sulfate	_____
60	nitrous acid	_____
61	dinitrogen tetroxide	_____
62	sodium bisulfite	_____
63	chlorous acid	_____
64	bromine trifluoride	_____
65	chlorine dioxide	_____
66	carbonic acid	_____
67	mercury (I) iodide	_____

- 68 titanium (II) oxide
- 69 hypochlorous acid
- 70 diiodine pentoxide
- 71 iron (III) chlorate
- 72 hydrocyanic acid
- 73 cadmium carbonate
- 74 silver phosphite
- 75 uranium (V) phosphate
- 76 sulfuric acid
- 77 silicon tetraiodide
- 78 silver sulfate
- 79 oxalic acid
- 80 magnesium hydride
